

Review of Sensors for In-Situ Amine Degradation Monitoring in Post-Combustion Carbon Capture

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Introduction

- \Box Reducing the CO₂ emissions is paramount to meet the decarbonization goal of net-zero emission by 2050
 - Post-combustion carbon capture offers a variety of advantages¹⁻⁴
 - 1. Retrofitted to existing coal fired power plants
 - 2. Suitable for natural gas fired power plants
 - 3. Power generation can be achieved even if the carbon capture process is down for maintenance unlike the pre-combustion process
- Chemical absorption is a widely used post-combustion method^{1–4}
- The most common chemical absorbers are amine-based solvents
 - 1. These solvent systems degrade losing their carbon capture efficiency over time
 - 2. Monoethanolamine (MEA) being the most studied
- Objective
 - In situ real-time monitoring of amine degradation will optimize operational control, carbon capture efficiency, and reduce the overall cost

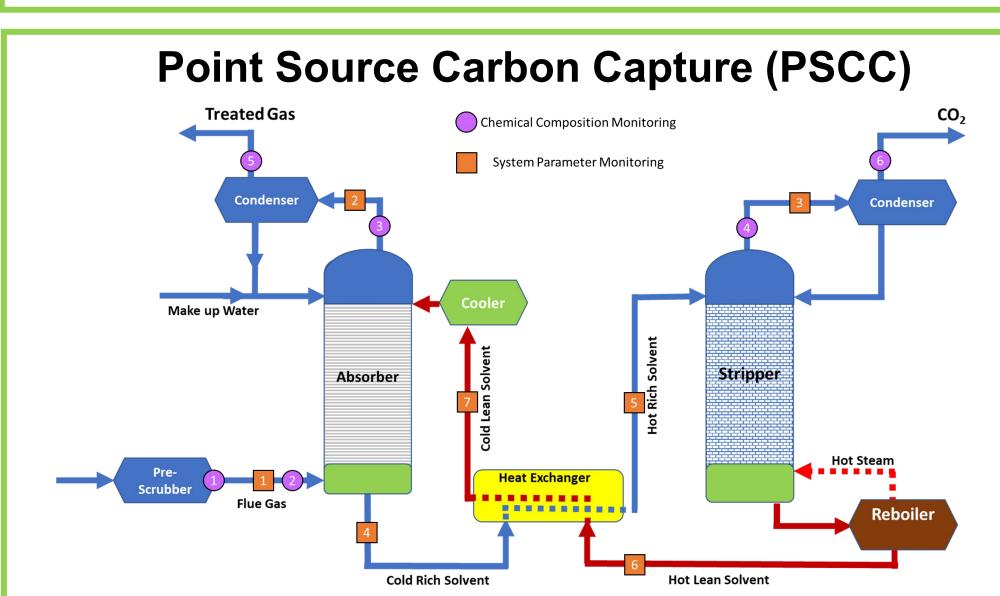


Figure 1. Pictorial representation of PSCC system with system parameter and chemical composition monitoring locations indicated

- **Amine Degradation Mechanisms**^{1,5-9}
- Oxidative: absorber, cross exchanger
- Thermal: stripper
- Caused by flue gas contaminants

Problem Statement: 1) Solvent degradation is hindering large-scale deployment of aminebased carbon capture. Amine solvent degradation associated costs can be significant compared with the cost to monitor. 2) Existing monitoring methods usually involve sampling from the process lines and sending samples to laboratories for analysis using expensive instruments.

References

- [1] Madejski, P.; et al., Energies (Basel) **15**(3) (2022).
- Control 72, 138-151 (2018).
- (2015). [4] Gouedard, C.; et al., International Journal of Greenhouse Gas [10] van Eckeveld, A. C., et al., Ind Eng Chem Res 53(13), 5515-
- Control 10, 244–270 (2012). [5] Lepaumier, H.; et al., Ind Eng Chem Res 48(20), 9061–9067 [11] Lv, B., et al., Environ Sci Technol 49(17), 10728–10735
- (2009). [6] da Silva, E. F.; et al., Ind Eng Chem Res 51(41), 13329–13338 (2012).
- [7] Ling, H.; et al., Sep Purif Technol **212**, 822–833 (2019).
- [2] Cuccia, L.; et al., International Journal of Greenhouse Gas [8] Morken, A. K., et al., Energy Procedia 114, 1245-1262, Elsevier Ltd (2017).
- [3] Dutcher, B.; et al., ACS Appl Mater Interfaces 7(4), 2137–2148 [9] Flø, N. E., et al., Energy Procedia 114, 1307–1324, Elsevier Ltd (2017)
 - 5523 (2014).
 - (2015).

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Industry Monitoring Necessities

Table 1. Industrial monitoring requirements both current and
 future

 Density of Solvent Viscosity of Solvent Temperature CO₂ Capture Efficiency Chemical Composition Pressure pH Amine Concentration PH Amine Concentration Water Mass Balance < 1 % Nitrosamine Concentration NH₃ Concentration in Water Wash SO₃ Impact on Aerosol Ammine Carryover Trace Metal: Mercury, Arsenic Selenium, Chromium pH Changes Following Distinct Functional Groups Color Change of Solvent Electrochemical Changes 	Current Monitoring Requirements	Emerging Monitoring Requirement
	 Viscosity of Solvent Temperature CO₂ Capture Efficiency Chemical Composition Pressure pH 	 < 1 % Nitrosamine Concentra NH₃ Concentration in Water Wash SO₃ Impact on Aerosol Ammine Carryover Trace Metal: Mercury, Arsenic Selenium, Chromium pH Changes Following Distinct Functional Groups Color Change of Solvent

Hardware Concerns

- Direct component analysis is prohibitively expensive
- Multiple sites need to be monitored continuously
- □ Self-contained systems are preferred due to hazardous solvent/contaminant
 - properties
- □ Hardware must be capable of running on multiple month timescale

Technology Gap

- □ Cost of analysis instrument
- Periodic sampling
- Point sensing
- □ Sensitivity to low-concentration degradation products
- □ Lack of monitoring of trace toxic metals

Table 2. Potential equipment cost for PSCC monitoring ¹⁰
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Equipment	Cost (\$)	
pH Meter/Automatic Titrator	\$3,000	
UV Gas Sensor	\$10,000	
Total Organic Carbon Analyzer	\$3,000	
Fourier-Transform Infrared	¢100.000	
Spectroscopy (FTIR)	\$100,000	
Nondispersive Infrared Sensor	\$20,000	
(NDIR)	φ20,000	
Paramagnetic O ₂ Analyzer	\$8,000	
Gas Chromatography–Mass	\$100,000	
Spectrometry (GC/MS)		
Liquid Chromatography–Mass	¢50,000	
Spectrometry (LC/MS)	\$50,000	
Electric Conductivity	\$1,000	
Single Ion Monitoring	< \$50,000	
	Aerosol	
Electric Low Pressure Impactor	Measurements (Size	
Electric Low-Pressure Impactor	Distribution and	
	Count)	

State of the Art Monitoring

Physical Parameters

Table 3. Physical monitoring parameters for PSCC²⁻⁶

ocation	Equipment	System Parameter Monitoring
1,2,3	Pressure Gauge	Pressure of Gas and Liquids
1,2	Volumetric Flow Rate	Rate of Gaseous Flow
4,5,6,7	Viscosity	Flow Rate of Solvent
4,5,6,7	Temperature	Temperature of Solvent

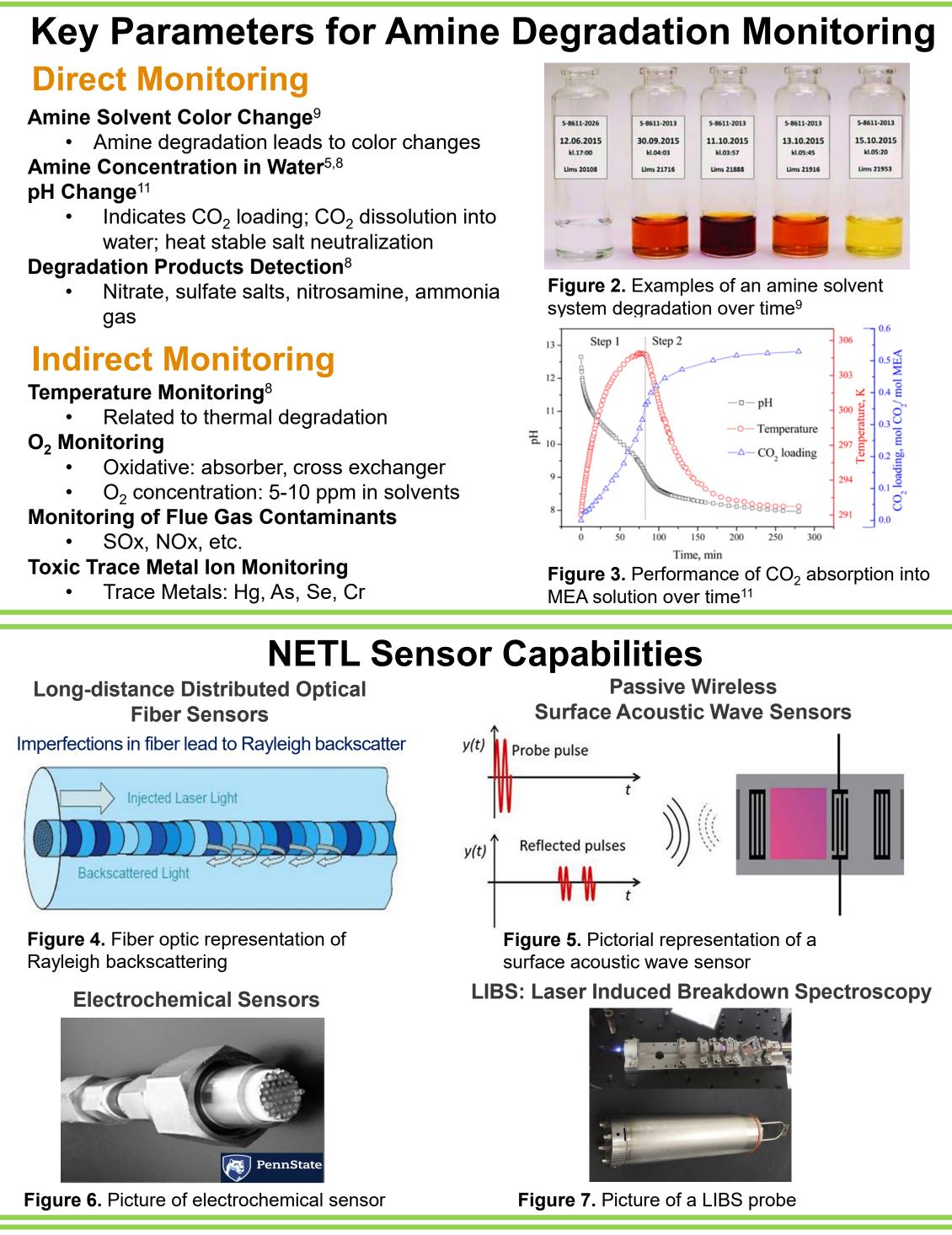
Monitoring locations for Table 2 and Table 3 are indicated in Figure 1.

Chemical Parameters

Table 4. Chemical monitoring parameters for PSCC²⁻⁶

Location	Equipment	Chemical Composition Monitoring
1	pH Meter	Basicity
1	UV	SO ₂ , NO ₂
1	Total Organic Carbon Analyzer	CO ₂
2,5,6	FTIR	CO_2 , H_2O , NH_3 , NO , NO_2 , SO_2 , CH_2O , C_2H_4O , Amines
2,5,6	NDIR	CO ₂
2	Paramagnetic	O ₂
3,4	GC/MS	CO ₂ , O ₂ , N ₂ , H ₂ O
3,4	LC/MS	CO ₂ , O ₂ , N ₂ , H ₂ O
2,4	Electric Conductivity	O ₂ content
5,6	Single Ion Monitoring	Mass Spectrometry
5,6	Electric Low-Pressure Impactor	Aerosol Measurements (Size Distribution and Count)





Summary

In-situ monitoring with NETL's sensor capabilities can be developed for deployment into the postcombustion carbon capture streams. These sensors will provide feedback on the carbon capture efficiency, solvent health, and reduce operational costs.

Project Achievement:

- 1. Identified key indicators for amine degradation as sensing targets.
- Surveyed and selected low-cost existing sensor technologies for these targeted indicators, instead of expensive full-on laboratory chemical analysis
- Interviewed industry stakeholders such as The National Carbon Capture Center (NCCC) and Ion Clean Energy to learn the monitoring needs for post-combustion carbon capture process.
- A report is prepared on monitoring needs, sensor technology survey, and recommendation for costeffective online monitoring of amine degradation

Next Step:

Pilot-scale testing of NETL-developed optical fiber sensors for amine degradation and CO₂ monitoring at NCCC.

