

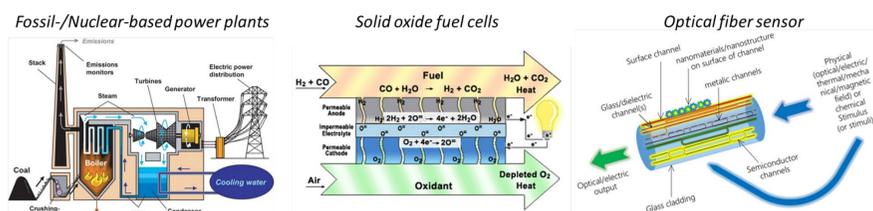
## Modeling and Experimental Testing of High-temperature Stable Sensor Materials for Gas Monitoring

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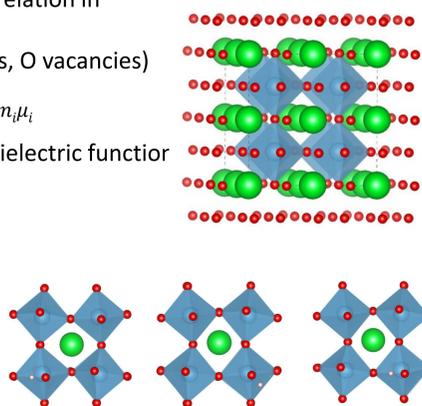
### Introduction

- For advanced real-time monitoring and control of gas species in combustion environments, development of efficient sensing platforms and materials under harsh environments are required
- Semiconducting optical-based sensor platform based on strontium titanate (SrTiO<sub>3</sub>) shows promise; cubic ABO<sub>3</sub> structure has tunable electronic, optical properties dependent on A- or B-site doping and density oxygen vacancies
- Hydrogen (H) and oxygen (O) impurities are thought to contribute to room temperature, long lasting photoconductivity in SrTiO<sub>3</sub>, indicating the potential use of SrTiO<sub>3</sub> in H and O gas sensing applications
- Understanding how lanthanum (La) and magnesium (Mg) doping of SrTiO<sub>3</sub> and oxygen vacancy defects affect electronic, optical properties of SrTiO<sub>3</sub> is required to tailor SrTiO<sub>3</sub>-based materials for development of sensitive, selective gas sensors
- First-principles modeling assesses the tunability of SrTiO<sub>3</sub> material properties via incorporation of La, Mg impurities and emergence of oxygen vacancies



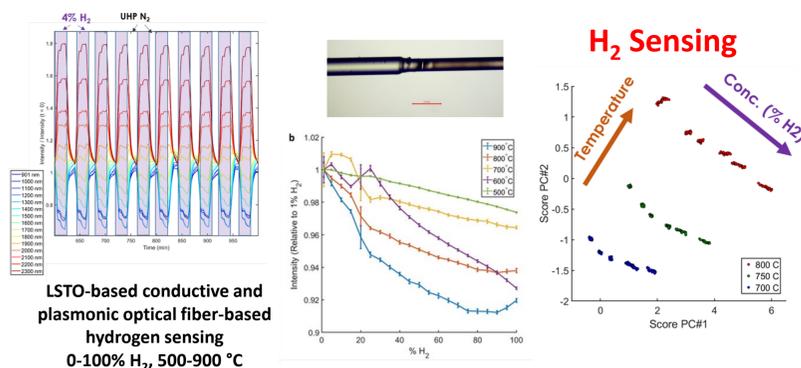
### Methods

- Density functional theory (DFT): PAW-PBE(+U) exchange-correlation in generalized gradient approximation (GGA)
  - Energies of formation of point defects (La and Mg dopants, O vacancies)
$$\Delta H(\text{SrTiO}_3, \text{def}) = E(\text{SrTiO}_3, \text{def}) - E(\text{SrTiO}_3) - \sum_i n_i \mu_i$$
  - Optical properties calculated from frequency-dependent dielectric function
- Incorporation of H and O impurities
  - Probing energetics of interstitial H and O atoms in SrTiO<sub>3</sub>
  - Locate local energy minima (binding sites) of H and O
  - Elucidate electronic, optical properties of SrTiO<sub>3</sub> with interstitial H and O atoms
  - Nudged elastic band calculations to determine relevant diffusion and associated energetic barriers



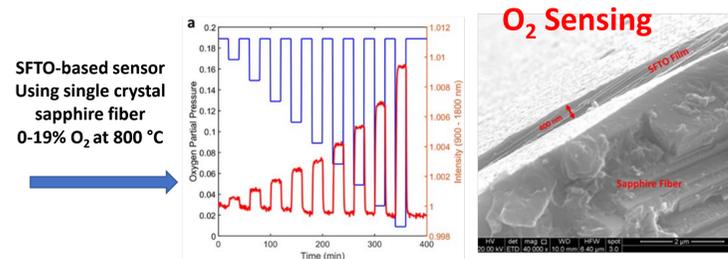
### Doped Perovskite Sensing Layers on Optical Fiber

- As a functional sensing layer on evanescent-field based optical fiber sensors, A- or B-site doped can operate as a versatile, high-temperature sensor for reducing or oxidizing gas streams
- La-doped SrTiO<sub>3</sub> acts like an n-type doped semiconductor under reducing conditions – demonstrated as an effective high-temperature sensing material for hydrogen



LSTO-based conductive and plasmonic optical fiber-based hydrogen sensing  
0-100% H<sub>2</sub>, 500-900 °C

- Other SrTiO<sub>3</sub>-based systems such as SrFe<sub>x</sub>Ti<sub>1-x</sub>O<sub>3</sub> (SFTO) and Mg-doped SrTiO<sub>3</sub> can act as p-type doped semiconductors under oxidizing conditions and show promise for high-temperature stable oxygen sensing



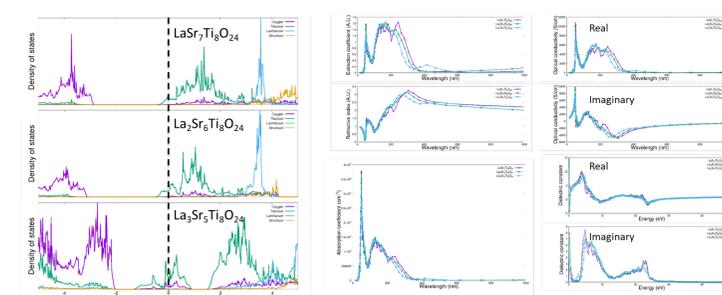
SFTO-based sensor Using single crystal sapphire fiber  
0-19% O<sub>2</sub> at 800 °C

### Disclaimer

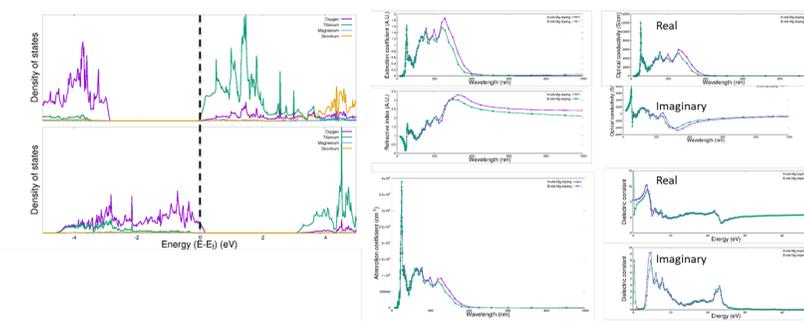
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### Perovskite Oxides: La- and Mg-doped SrTiO<sub>3</sub>

- La-doped SrTiO<sub>3</sub> shows n type behavior for all La doping levels studied ranging from 12.5 to 37.5 at.%; La doping impacts the free carrier concentration
- Imaginary component of dielectric matrix at sub-bandgap energy of SrTiO<sub>3</sub> confirms metallicity; La doping causes shift in dielectric matrix to higher photon energy above 4 eV
- La doping causes drop in NUV absorption due to shifting of band edge



- Mg-doped SrTiO<sub>3</sub> exhibits changes of electronic, optical properties dependent on site of Mg substitution; neither show absorption in visible light range
- B-site doped SrTiO<sub>3</sub> has occurrence of Drude peak in imaginary component of dielectric matrix, confirming the presence of free carriers



### Publications

- J. Park *et al.*, *Phys. Chem. Chem. Phys.* **22**(2020) 27163-72; *ACS Appl. Mater. Interfaces* **13**(2021) 17717-25; *J. Phys. Chem. C* **125**(2021) 22231-38; **126**(2022)8832-38; *Chem. Mater.* **34**(2022)6108-15
- Y.-N. Wu *et al.*, *J. Phys. Chem. C* **122**(2018) 22642-49; *J. Phys. Chem. Lett.* **11**(2020) 2518-23; *J. Phys. Chem. Mater.* **32**(2020) 405705.
- T. Jia *et al.*, *RSC Adv.* **7**(2017) 38798-804; *Phys. Chem. Chem. Phys.* **22**(2020) 16721-26; *Applied Energy* **281** (2021)116040; *J. Phys. Chem. C* **125**(2021) 12374-81; **126**(2022)11421-25
- Y. Duan *et al.*, *J. Solid State Chem.* **256**(2017) 239-251.
- S. Nations, *et al.*, *RSC Adv.* **11**(2021) 22264-72; *Mater. Adv.* **3**(2022)3897-3905; *Nanomaterials* **13**(2023)276