

## Pd/PMMA nanocomposite coated fiber optic hydrogen sensor

Tulika Khanikar, Dolendra Karki, Yang-Duan Su, Jun Young Hong, Yuankang Wang, Khurram Naeem and Paul R. Ohodnicki  
 Department of Mechanical Engineering and Materials Science, University of Pittsburgh, PA, USA.

### Material Characterization

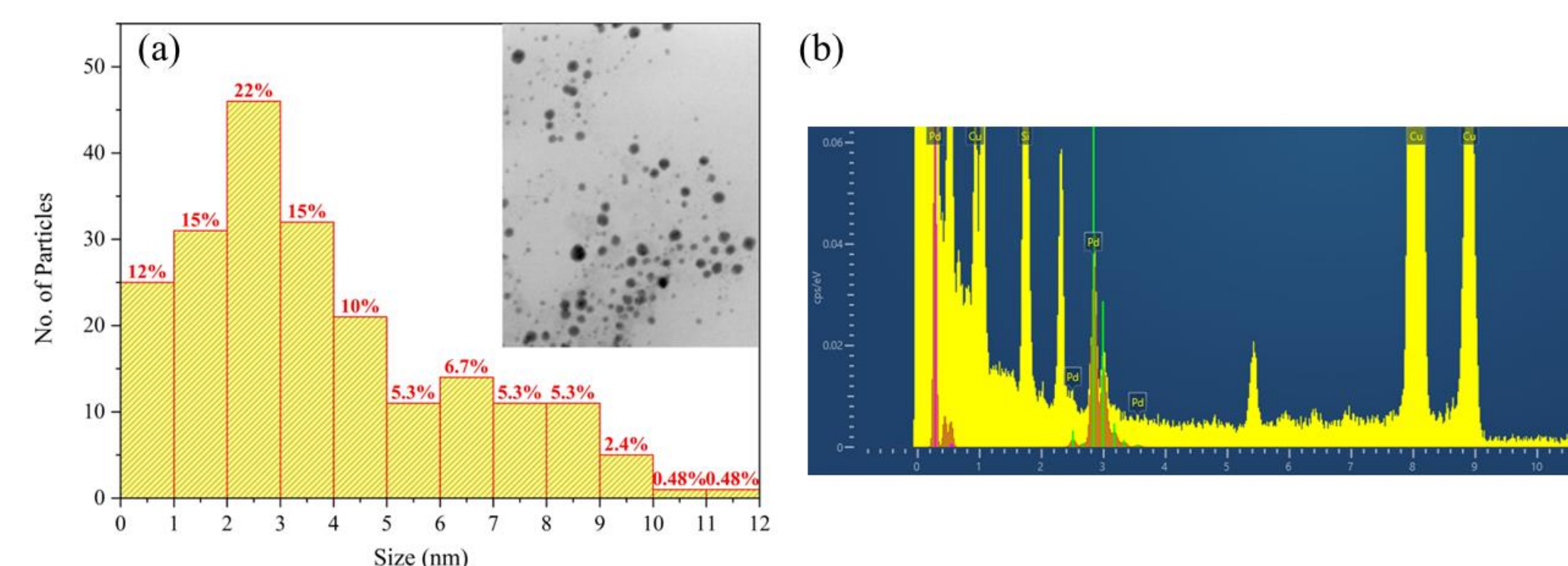


Fig. 1 (a) Particle size distribution analysis (b) STEM-EDS mapping of segregated Pd NPs

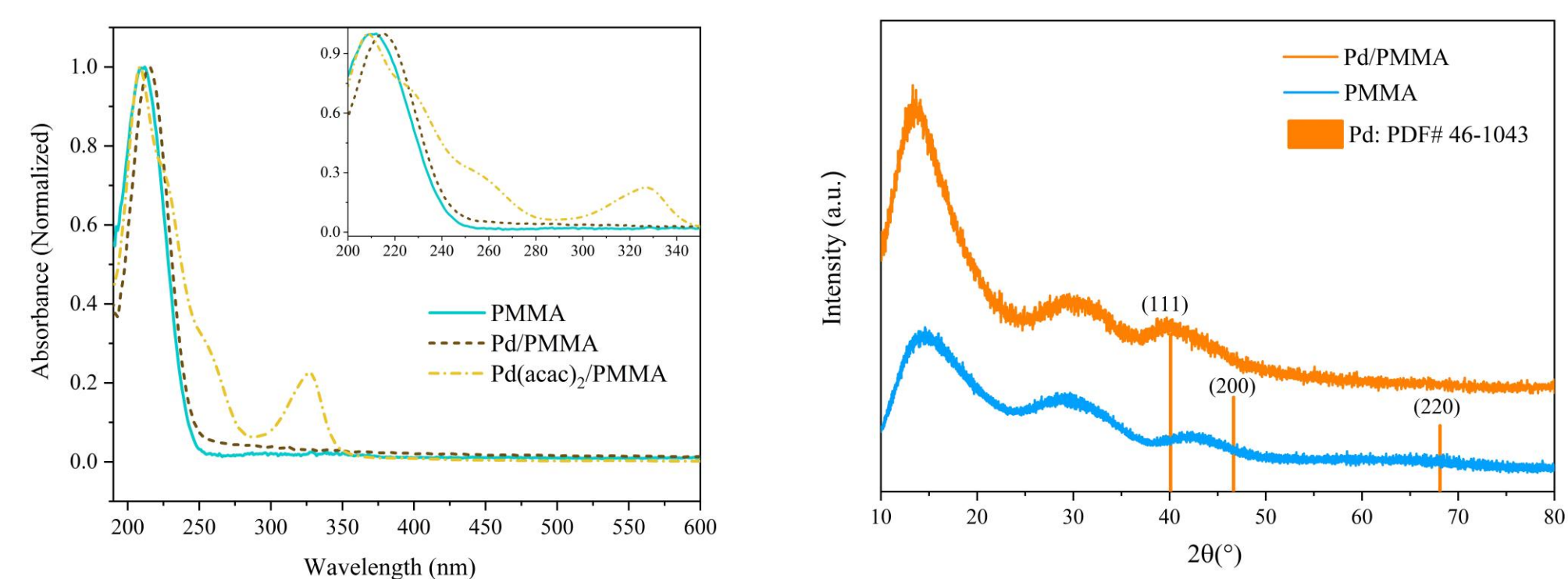


Fig. 2. UV-VIS spectroscopic analysis (left), GI-XRD spectra of the composite material (right)



Fig. 3. FESEM images of coated fiber surface (a), cross section (b), and zoomed view cross section

- Microscopic analysis of the nanocomposite indicate the existence of well dispersed ultrafine Pd NPs without large aggregates, most particles (74%) are less than 5 nm in size
- The coated fiber surface appears smooth and homogeneous with no particulates. The cross-sectional image indicates conformal coating with thickness of  $\sim 2.8 \mu\text{m}$

### Overview and Highlights

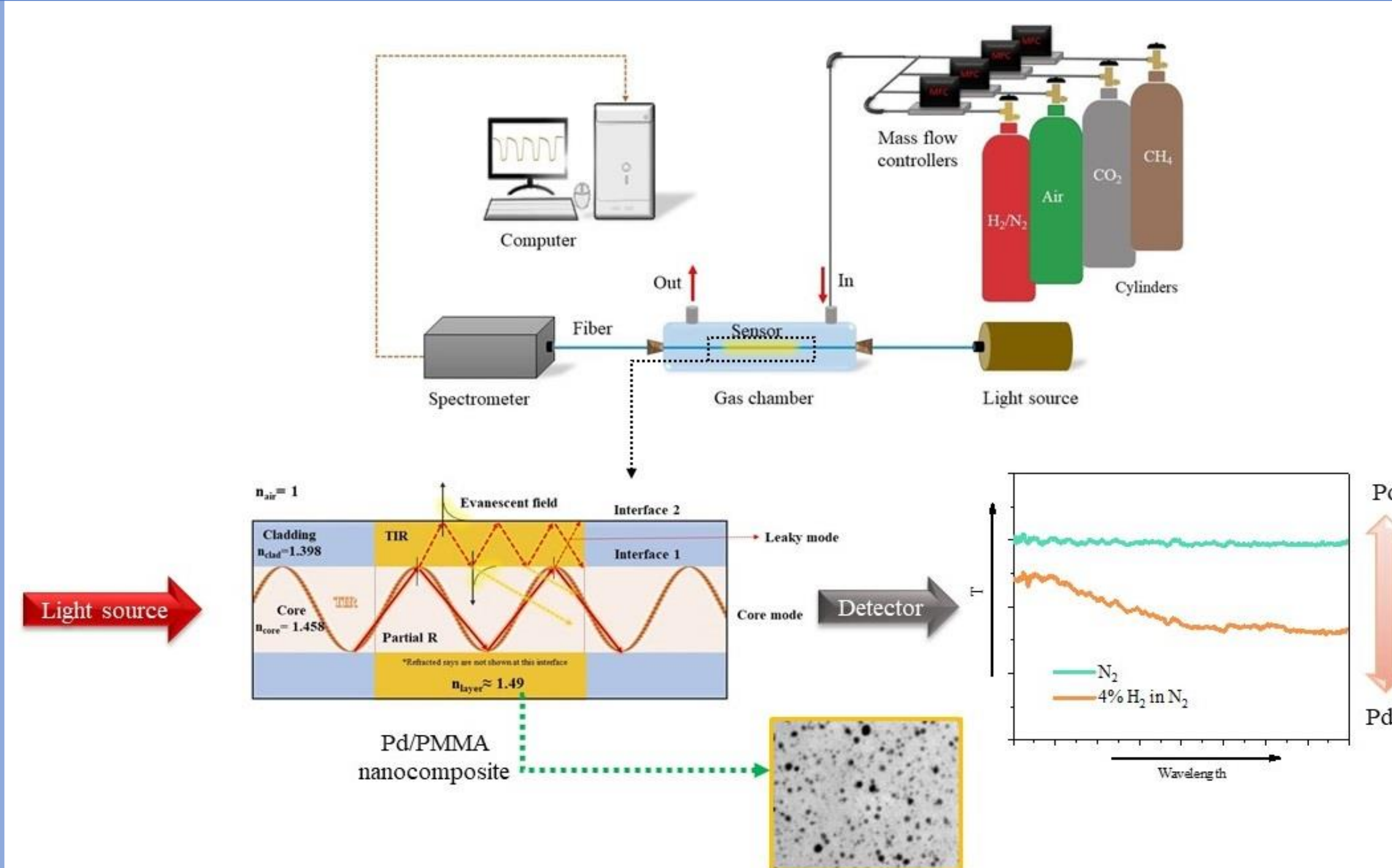


Fig. 4. Graphical abstract

- A fiber optic hydrogen sensor coated with a palladium/polymethyl methacrylate (Pd/PMMA) nanocomposite layer is fabricated and benchmarked here.
- The sensor with Pd/PMMA substituted cladding, operates by intensity modulation technique through partial reflection in the sensing region.
- In-situ thermal reduction method has been adopted for nanocomposite synthesis due to reduced processing steps required and to avoid material waste.
- The feasible and cost-effective sensor preparation protocol make the sensor a competitive candidate for hydrogen gas monitoring applications that operate at near room temperatures.

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### Fiber optic sensor Characterization

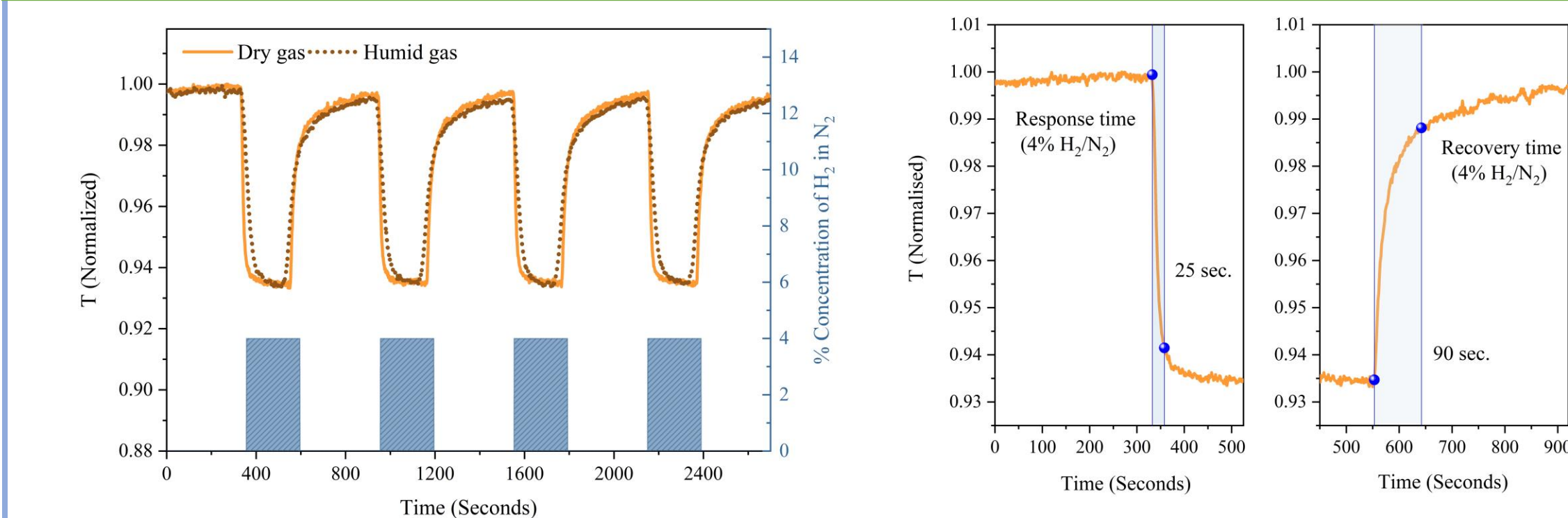


Fig. 5. (L) Response to 4%  $\text{H}_2/\text{N}_2$  (R) Response and recovery time towards 4%  $\text{H}_2/\text{N}_2$  under dry gas condition

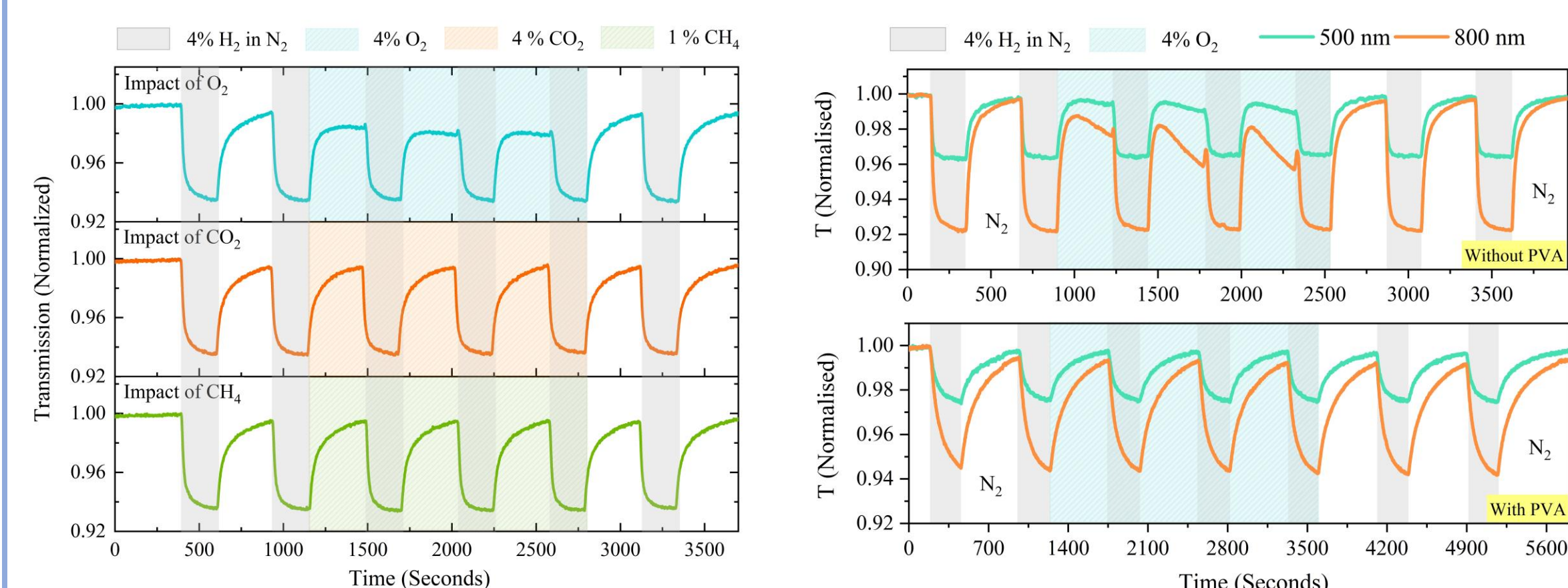


Fig. 6. (L) Sensor response in presence of interfering gases (R) Before and after coating with secondary PVA layer in response to changing gas composition

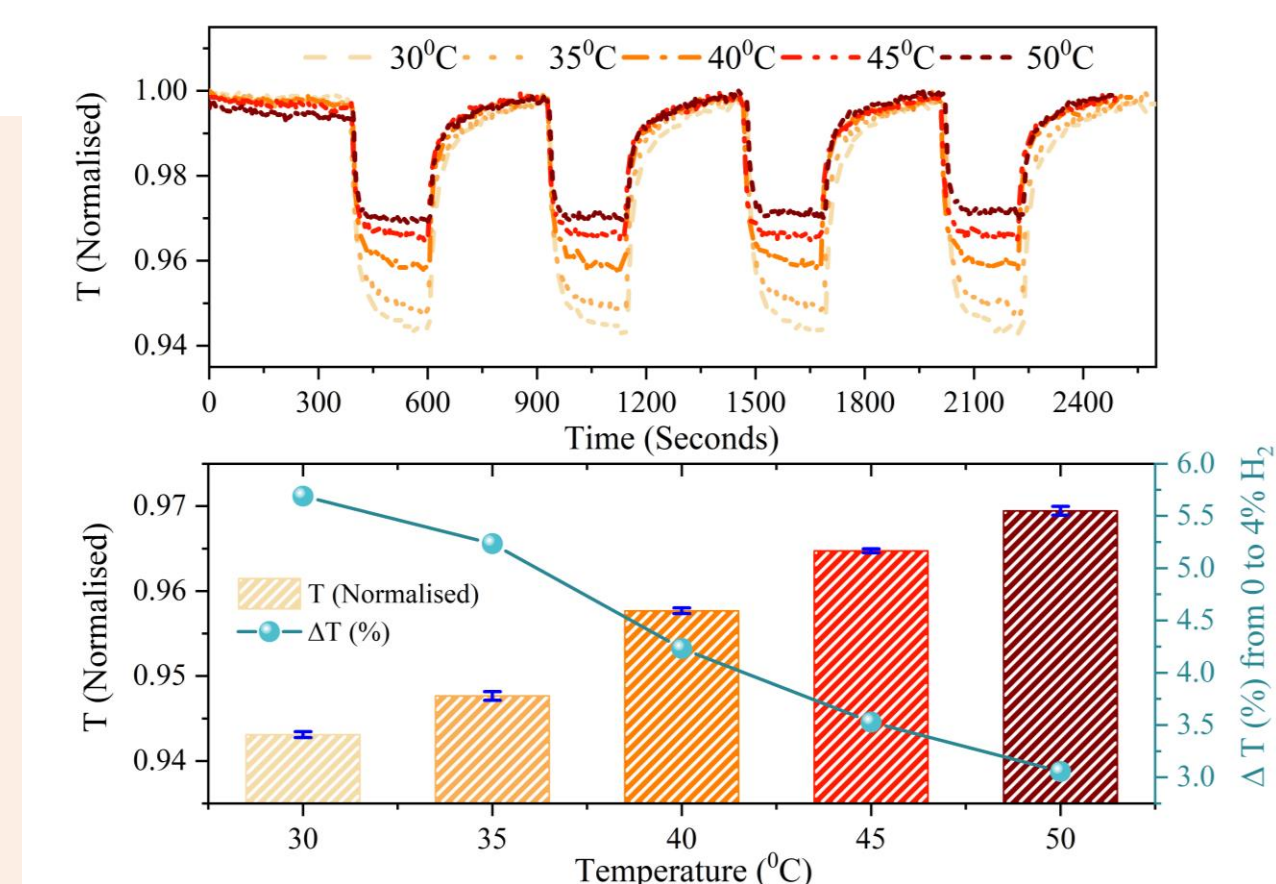


Fig. 7.  $\text{H}_2$  sensing response (4%) at different temperatures

- Excellent humidity resistance
- Good environmental stability in presence of interfering gases.
- Response time of 25 sec to 4%  $\text{H}_2$ .
- Can detect low  $\text{H}_2$  concentration, down-to 0.1%

However, presence of  $\text{O}_2$  and increased temperature contributes to lowering the sensing response which could again be restored once the sensor is moved to inert environment at room temperature.

### Acknowledgement

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